

Research Journal of Pharmaceutical, Biological and Chemical Sciences

Flavonoids from Anthemis stiparum Pomel.

S Benferdjallah ^{1,2*}, H Dendougui ^{1,2}, F Benayache ¹, S Benayache ¹, F León ³, I Brouard³, JC Hernándezand, and J Bermejo³

¹Laboratoire de Valorisation des Ressources Naturelles, Université Mentouri Constantine, Algeria ²Laboratoire de valorisation et promotion des ressources sahariennes (VPRS), Université Ouargla, 30 000 Ouargla, Algeria ³Instituto de Productos Naturales y Agrobiologia-C.S.I.C., Instituto Universitario de Bio-Orgánica, "Antonio González", Av. Astrofísico F. Sánchez 3,38206 La laguna, Tenerife, Spain

ABSTRACT

Three flavonoid aglycones ,Hispidulin (4',5,7- trihydroxy-6-methoxyflavone), Jaceosidin (4',5,7-trihydroxy-3', 6-dimethoxyflavone) and Eupatilin (5,7- dihydroxy-3',4', 6-trimethoxyflavone) have been identified for the first time from the aerial parts of *AnthemisStiparumPomel*. The structures of compounds were determined using detailed spectroscopic analyses.

Keywords: Asteraceae, Anthemis stiparum, Flavonoids.

*Corresponding author



INTRODUCTION

Flavonoids are natural products found as biochemical constituents of plants, Because of their potential therapeutic significance [1], the number of identified flavonoids is increasing rapidly and extensive screening of their actions is being carried out in many laboratories [2-4]. Anthemisis a genus of about 130 species of aromatics herbs in the Asteraceae [5], is grown in Algeria in about 10species.Specie*Stiparum* is an endemic plant in Algeria and Morocco [6]. Several members of this genus are used in folk medicine. Algerians use *Anthemis stiparum* as an Antispasmodic. The phytochemical studies of this plant have not been reported. We have now investigated the chemical constituents of this plant. The chromatographic separation of the ethyl acetate extracts of a concentrated ethanol from *Anthemis stiparum* afforded three known flavones aglycones, Hispidulin (I), Jaceosidin (II) and Eupatilin (III).

EXPERIMENTAL

a) Experimental procedure.

NMR spectra were obtained on a Bruker model AMX-400 spectrometer with standard pulse sequences, operating at 400 MHz for ¹H. CDCl₃ and (CD₃)₂CO was used as solvent and TMS as internal standard. EIMS were taken on a Micromass model Autospec (70 eV) spectrometer. UV Spectra (MeOH): Shimadzu (190–3200 nm, UV-3101PC) spectrophotometer. Column chromatography (CC) was carried out with Silica gel 60 (Merck, 200-400 mesh). Flash column with Silica gel fine (Merck, Silica gel 60 PF254). TLC: Silica gel 60(Merck, F254) and the detection with a spraying reagent (CH₃COOH/H₂O/H₂SO₄; 80:16:4) followed by heating.

b) Plant material.

ANTHEMIS STIPARUM POMEL was collected from Stil area (El-oued), in south of Algeria in April 2005 and the plant was identified by Dr.YoucefHallis on the basis of Quezel and Santa[6] A voucher specimen has been deposited in the Herbarium of the laboratory of biochemistry, scientific and technical center for arid areas (CRSTRA), Touggourt, Algeria under the code number Astantmsti2005

c) Extraction and isolation.

Air dried aerial parts of *Anthemis stiparum* (3.42Kg)were extracted three times with EtOH at room temperature, the EtOHextracted was evaporated in vacuo to give dark brown syrup (324g).The syrup was suspended in water and extracted three times with EtOAc.The ethyl acetate layer was concentrated in vacuo to give a black syrup(158g) which was subjected to chromatography on silica gel and eluted with a gradient of n-hexane – EtOAc and EtOAc-MeOH to give 23 fractions(23f).300 mg of f15(n-hexane – EtOAc 1:1)was subjected to chromatography on silica gel and eluted with CHCl₃-Acetone to give I(11mg),III(12mg),III (21mg).

Hispidulin (I):

Pale yellow; EIMS and UV see tables(1,2); HNMR (400MHz, (CD₃)₂CO): δ = 3.74ppm (3H, s, 6-OMe), δ =6.49ppm (1H, s, H-8orH-3) δ =6.50ppm (1H, s, H-3or H-8) , δ =6,89ppm(2H,d,J=8.6Hz,H-3', 5'), δ =7,83ppm (2H,d,J=8.56;H-2', 6'), δ =9.1ppm (1H, 5OH).

Jaceosidin (II):

Pale yellow ; EIMS and UV see tables(1,2) ; HNMR(400MHz,(CD₃)₂CO) δ =3.74 and 3.86ppm (6H,s,two-OMe), δ =6.50ppm (1H, s, H- 3or H-8) and δ =6.57ppm (1H, s, H-3or H-8), δ =6.87ppm(1H,d,j=8.32Hz, H-5') , δ = 7.48ppm(1H,dd,j=8.40,1.84Hz, H-6'), δ = 7.51ppm(1H,d,j=1.84Hz,H-2').

Eupatilin (III):

Yellow; EIMS and UV see tables(1,2) ; HNMR(400MHz,(CDCl₃) δ =3.89,3.90,3.97ppm (9H,s,three-OMe), δ = 6.90ppm(1H,d,j=8.4Hz,H-5'), δ =7.19ppm (1H, s, H-2'), δ =7.44ppm (1H, d, J=8.4Hz, H-6').

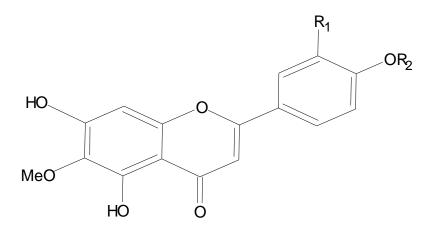


RESULTS AND DISCUSSION

Compound I was obtained as a pale yellow. The EIMS of I exhibited a molecular ion peak at m/z 300(100%) for C₁₆H₁₂O₆ in accord with a flavones containing three hydroxyl and one methoxyl group, the peak m/z=69is diagnostic of 6-oxygenated [7],and the peak (M-43) ⁺ as characteristic of 6-methoxy flavones and flavanols(table1)[8] .I appeared as purple fluorescent spot on a paper chromatogram in UV light, changing to yellow with ammonia indicating the presence of free 5-and 4'-hydroxyl groups .UV maxima in methanol at 336 and 276 and the shifts obtained with diagnostic reagents(table2) suggested the presence of a methoxyl group at C-6(Δ +25nm for band I in AlCl₃/HCl compared to band I in MeOH) and hydroxyl groups at positions 5,7,4'[9-12].The ¹HNMR spectrum of I exhibited one singlet at δ 3.74 for one methoxy group, singlets at δ 6.49 and6.50 assigned to H-3 and H-8 or H-8 and H-3respectively,and a two proton doublet appeared at δ 6,89 which is acharacteristic of H-3' and H-5',and a two proton doublet appeared at δ 7,83 which isa characteristic ofH-2' and H-6'and signal for one proton at 9.1 for 5-OH[13-14].these spectral data established the structure of I as 4',5,7-trihydroxy-6-methoxyflavone (hispidulin).

Compound II was obtained as a pale yellow. The EIMS of II exhibited a molecular ion peak at m/z 330(100%) for C₁₇H₁₄O₇ in accord with a flavones containing three hydroxyl and two methoxyl groups, the peak m/z=69is diagnostic of 6-oxygenated[7], and the peak (M-43) ⁺ as characteristic of 6-methoxy flavones and flavanols (table1) [8] **.II** appeared as purple fluorescent spot on a chromatogram paper in UV light, changing to yellow with ammonia indicating the presence of free 5-and 4'-hydroxyl groups .UV maxima in methanol at 344 and 275 and the shifts obtained with diagnostic reagents(table2) suggested the presence of a methoxyl group at C-6(Δ +14 nm for band I in AlCl₃/HCl compared to band I in MeOH) and hydroxyl groups at positions 5,7,4' [9-12].The ¹HNMR spectrum of II exhibited two singlets at δ 3.74 and 3.86 for two methoxy groups, singlets at δ 6.50 and 6.57 assigned to H-3 and H-8 or H-8 and H-3respectively,and doublet at δ 6.87assigned to H-5',and doublet-doublet at δ 7,48 assigned to H-6',and doublet at δ 7,51 assigned at H-2'[14]. these spectral data established the structure of II as 4',5,7-trihydroxy-3', 6-dimethoxyflavone(Jaceosidin).

Compound III was obtained as a yellow. The EIMS of III exhibited a molecular ion peak at m/z 344(100%) for C₁₈H₁₆O₇ in accord with a flavones containing two hydroxyl and three methoxyl groups, the peak m/z=69is diagnostic of 6-oxygenated [7], and the peak (M-43) ⁺ as characteristic of 6-methoxy flavones and flavanols (table1) [8]. The UV maxima in methanol at 342 and 276 and the shifts obtained with diagnostic reagents(table2) suggested the presence of a methoxyl groups at C-6(Δ +18 nm for band I in AlCl₃/HCl compared to band I in MeOH) and hydroxyl groups at positions 5,7 and the absent of 4'-OH[9-12].The ¹HNMR spectrum of III exhibited three singlets δ 3.89,3.90,3.97 for Three methoxy groups, doublet at δ 6.90 assigned to H-5', and singlet at δ 7.19 assigned to H-2',and doublet at δ 7.44 assigned to H-6'[13-14].. These spectral data established the structure of III as 5, 7- dihydroxy-3', 4', 6-trimethoxyflavone (Eupatilin).



I:R1=H,R2=H; II:R1=OMe,R2=OME; III:R1=OMe, R2=Me



Table 1: MS data for flavonoids												
Flavonoid	[M]⁺	[M-H]⁺	[M-Me]⁺	[M-H₂O]⁺	[M-HCO]⁺	[M- COMe]⁺	[A₁- Me]⁺	[A₁- MeCO]⁺	[A ₁ -MeCO-CO] ⁺	[B₁]⁺	[B ₂]+	[(CH2)₃CH=CH2] ⁺
Hispidulin I	300	299	285	282	271	257	167	139	111	118	121	69
(%)	(100)	(7)	(68)	(39)	(8)	(56)	(15)	(16)	(1)	(11)	(6)	(46)
Jaceosidin II	330	329	315	318	301	287	167	139	111	148	151	69
(%)	(100)	(7)	(66)	(40)	(6)	(40)	(10)	(9)	(1)	(4)	(4)	(28)
Eupatilin III	344	343	329	326	315	301	167	139	111	162	165	69
(%)	(100)	(6)	(66)	(48)	(7)	(39)	(7)	(8)	(~~0)	(4)	(6)	(13)

MS were recorded at 70eV.Valuesare given in m/z and in parentheses the % abundance relative to the base peak. The A₁, B₁, and B₂ terminology for the fragments is given in [12].

	Table 2: UV data for flavonoids									
Flavonoid	MeOH(λmax, nm)	NaOH(λmax, nm)	AlCl3(λmax, nm)	AlCl3/HCl(λmax, nm)	NaOAc(λmax, nm)	NaOAc/H3BO3(λmax, nm)				
	336	396	362	361	380	348				
Hispidulin I	276	328	303	292	277	276				
		276	282	260						
	344	408	374	360	399	361				
Jaceosidin II	275	337	282	281	322	277				
		275	262	260	276					
	342	370	368	359	369	351				
Eupatilin III	276	309	287	290	314	276				
		275	268	257	276					



ACKNOWLEDGMENTS

The authors wish to thank Dr.Youcef Hallis (laboratory of biochemistry, scientific and technical center for arid areas, Touggourt, Algeria) for the identification of the plant material.

REFERENCES

- [1] HarborneJB , WilliamsCA.Phytochemistry2000; 55(6):481-504.
- [2] TeilletF, BoumendjelA, BoutonnatJ,RonotX .MedResRev2008;28(5) :715-45.
- [3] AbdallaS, Abu-ZargaM, Al-KhalilS, SabriS .General Pharmacology1988;19(4):559-63.
- [4] GiangasperoA, PontiC, PollastroF, Del FaveroG, Della LoggiaR, TubaroA, AppendinoG, SosaS.J Agric Food Chem2009; 57 (17):7726-30.
- [5] Teixeira da Silva J A .Afr J Biotech2004; 3(12): 706-720.
- [6] QuezelP, SantaS .Nouvelle flore de l'Algérie et des régions désertiques et méridionales. CNRS,Paris , 1963; 973.
- [7] SchmidtTJ, MerfortI. Journal of Chromatography1993; 634 : 350-55.
- [8] Bomfim-PatrícioMC, SalatinoA, Martins B, SalatinoM LF,WurdackJJ.Biochemi System and Ecolo2001; 29(7): 711-26.
- [9] Mabry TJ, MarkhamKR, ThomasMB.The systematic identification of flavonoids.Springer-Verlag, First edition,Berlin, 1970; 41-61.
- [10] Geissman A.The chemistry of flavanoid compounds.the MacMillan Company, New York, 1962; 72.
- [11] VoirinB.phytochem1983; 22(10):2107-45.
- [12] harboneJB,MarbyTJ,MarbyH.The flavonoids, academic press, new york,1975, 46-126.
- [13] AkkalS , BanayacheF, BentameneA, MedjroubiK, SeguinE, TillequinF.Chem Nat Comp2003; 39(2):219-20.
- [14] Yan-Ru D, Ai-Xin S, Han-Qing W. JChiChemSoc 2004; 51 (3): 629-36.